Local Motifs and the Rise of Solidity in Deeply Supercooled Liquids





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Introduction. We investigate the role that Locally Favoured Structures (LFS) play in the emergence of the solid-like behaviour of glassy systems in numerical simulations. To do so, we start from extremely cold configurations (T~0.97Tg) and perform a shear experiment testing the zero-temperature limit of a set of rheological properties. We distinguish the more plastic and the more elastic regions and correlate them with the localisation of the LFS.

1. Generating highly structural configurations via importance sampling of trajectories

Locally Favoured Structures (LFS) in supercooled liquids are local particle motifs that are energetically favourable and linked to transient short range order.

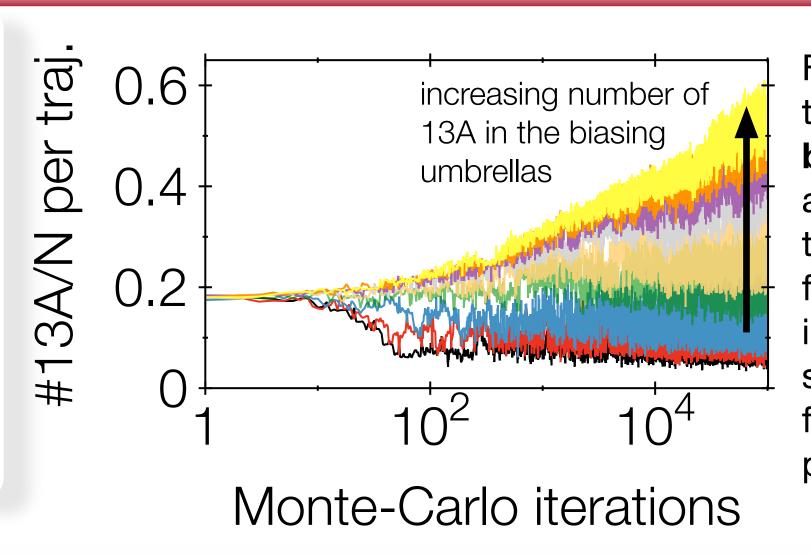


Fig. 1 Sampling trajectories biased according to the time-integrated fraction of icosahedral structures 13A for N=512 particles.

We consider a simple binary mixture [1] where icosahedral LFS have been observed. We manage to obtain high concentrations of icosahedra (13A) via **Transition Path Sam**pling (TPS) [2].

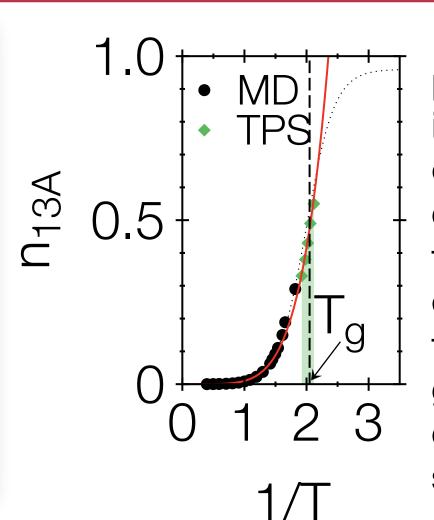
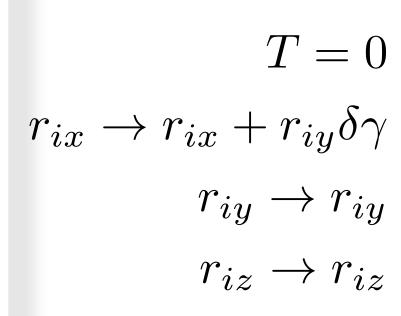


Fig. 2 LFS-rich phases can be interpreted as very cold samples. The corresponding temperature can be extrapolated from the icosahedra growth law determined via standard Molecular Dynamcis [2].

2. Probing the mechanical response in the Athermal Quasistatic Limit

The relation between solidity and icosahedral order is probed through a rheological test: the sample is sheared at T=0 and performs a succession of

affine deformations (at constant shear rate) and potential energy minimisations [3].



Eq.1 The equations of motion are fully deterministic with a constant shear rate $\dot{\gamma}$.

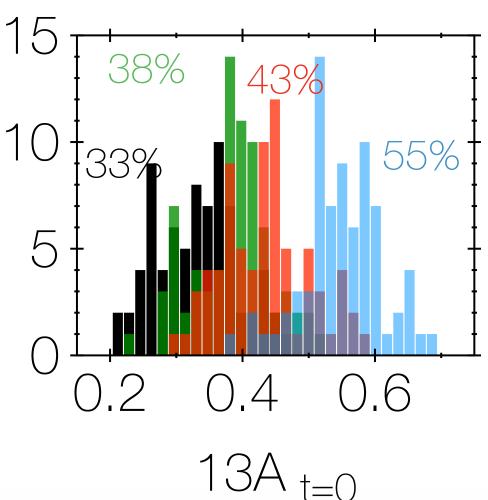
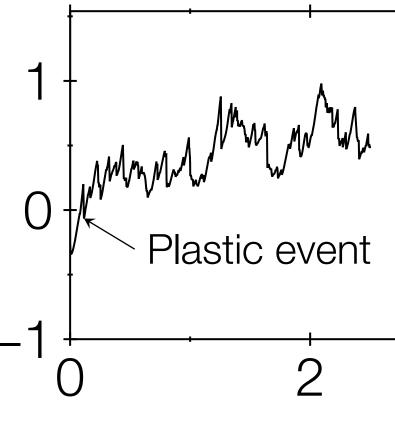


Fig. 3 The TPS umbrellas provide consistent **ensembles** of \supset configurations characterised by large values of the concentration of icosahedra.



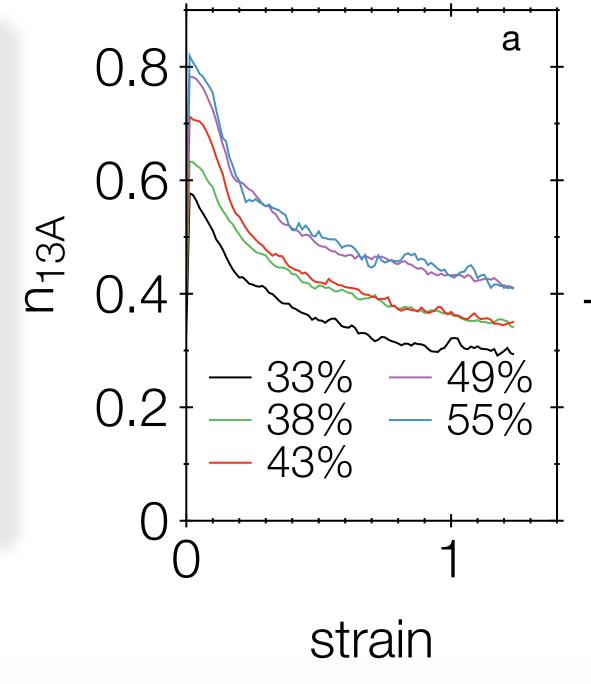
strain

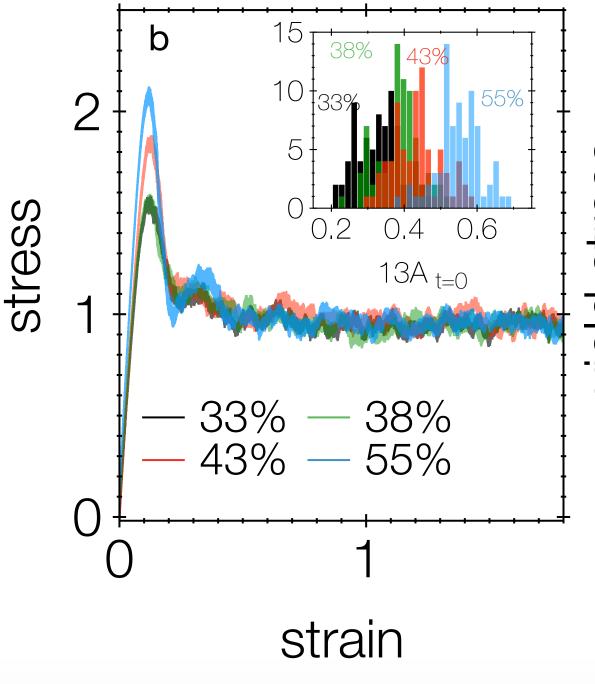
Fig. 4 The potential energy for different values of the strain illustrates the succession of plastic events.

3. More icosahedral structures enhance the yield stress and the elastic modulus

While the energy minimisation initially optimises the fraction of 13A, the shearing protocol leads to a gradual fluidification.

This is **strongly** dependent on the initial fraction of icosahedra.





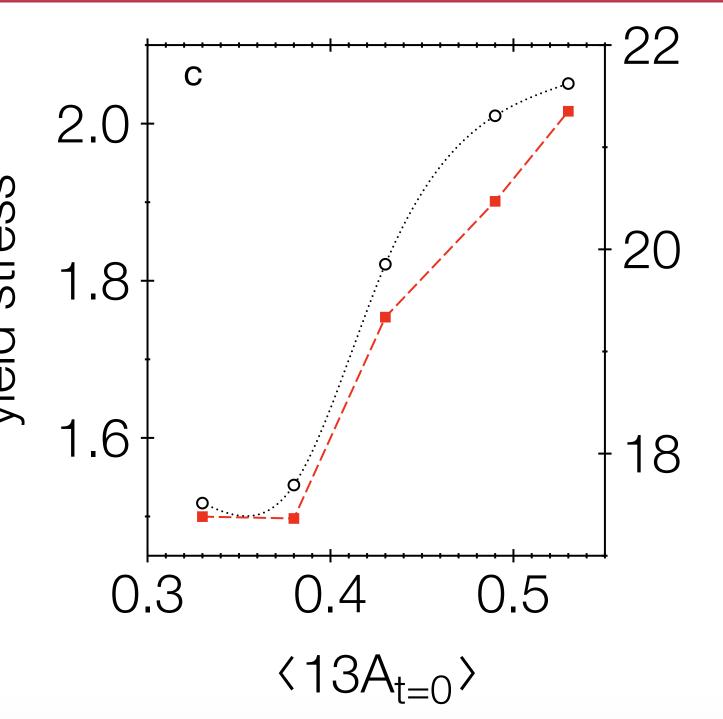
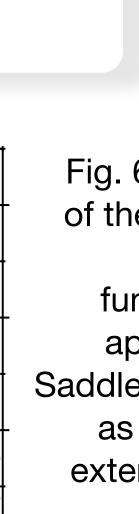
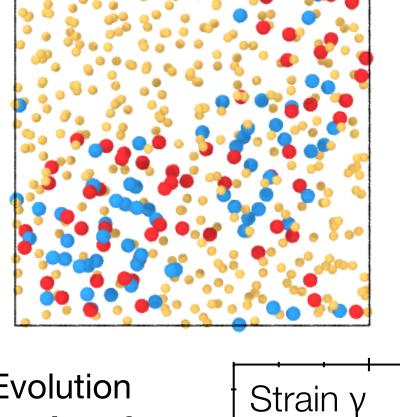


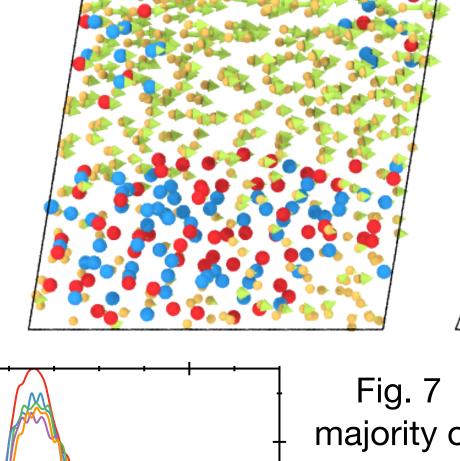
Fig. 5 Fluidification of the the initial configurations: a) decrease of the fraction of icosahedral particles; b) stress-strain curves and c) related measurements of yield stress (stress-peak) and Young's modulus for different initial average concentrations of LFS.

4. The plastic events occur outside the icosahedral clusters

Dynamic information from the **non-affine** displacements [4] and static features from the mode analysis [5] anti-correlate with the location of the icosahedral clusters.







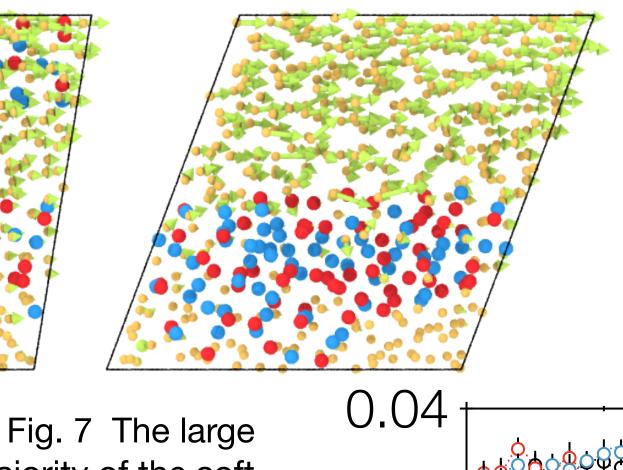


Fig. 8 Displacement field and icosahedral order: shearbanding can be observed, with much larger mobility in the regions not occupied by the LFS structure.

the averages of Dmin, the local deviation from uniform strain [4] between icosahedra and the rest of the sample.

Fig. 10 The plastic events

in between the icosahedral

clusters (in red and blue).

determined from the 7 softest

modes) are localised outside and

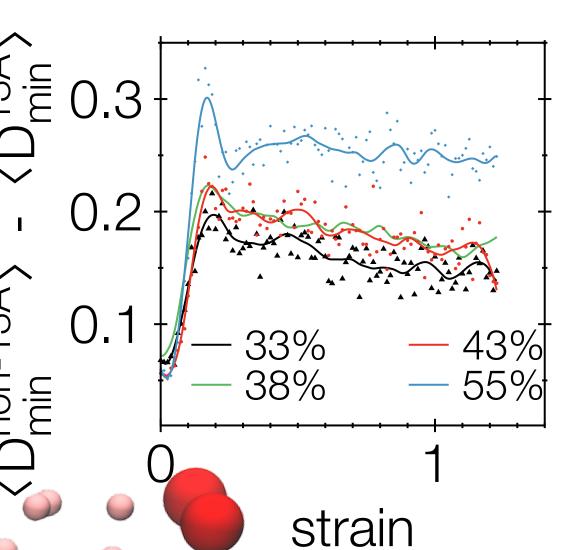
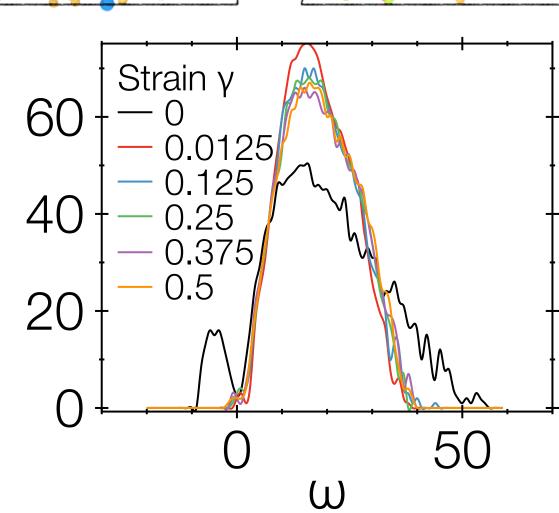
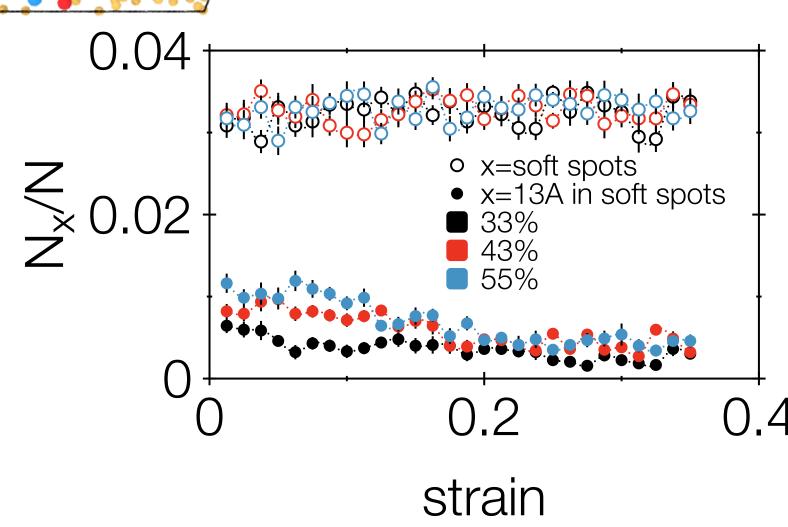


Fig. 6 Evolution of the **density of** 60 states as a function of the 40 applied strain. Saddles disappear as soon as the 20 + external shear is applied.



majority of the soft spots is not localised into icosahedral regions, even when these occupy the most of the volume.



Conclusions. Using biased sampling techniques, we obtained **high concentrations** of LFS. These appear to be correlated with the **hardening** of the rheological response.

The **localisation** of the LFS is **anti-correlated** with non-affine **displacements** and mobilitytriggering **soft-spots**, justifying the interpretation of the icosahedral LFS as the **backbone** of the solid structure of the considered glass former.

References & Acknowledgements

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